DEPARTMENT OF CHEMISTRY

GRADUATE COURSE IN MASS SPECTROMETRY: LECTURE 8

Mass spectrometry for biophysics and structural biology





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Levels of protein structure



Mass spectrometry can inform on all levels of protein structure

What do we want to know?

- Structure
 - Stoichiometry, connectivity, 3D topology, conformations, atomic coordinates
- Thermodynamics
 - Interface and interaction strengths, stability
- Kinetics
 - Rate constants, activation energies



Two strategies - native or labelling



Labelling technologies with MS

- Label protein, and use MS to localise the individual labels
- Can interrogate the 'global' and 'local' levels: i.e. protein or peptide level
- Peptide level accessed by enzymolysis of protein, and further interrogation with tandem MS
- Labels reveal solvent accessibility and connectivity
- Provide means of probing protein structure and dynamic fluctuations



Cross-linking mass spectrometry



- Cross-linker forms covalent bonds between amino acids with appropriate functionality
- Peptide masses (and fragments) are interrogated to localise cross-linker
- Intra- and inter-protein cross-links can be formed
- Cross-links can be used to determine connectivity, and as a spatial restraint for modelling protein structures
- Problems centre on decreased detection efficiency of cross-linked peptides, kinetic effects, and interpretation of observed links

Oxidative foot-printing



- · Solvent accessible amino acid side chains of the protein are oxidised
- Peptide masses (and fragments) are interrogated to localise oxidation sites
- Comparing data from proteins in complex and in isolation allows the determination of interface sites
- Problems centre on data analysis and differential intrinsic reactivity

Hydrogen/deuterium exchange

- Monitor the rate at which protein hydrogens are replaced by deuteriums (or vice versa)
- Three types of hydrogen in proteins, only backbone amide hydrogens exchange at measurable rate
- Exchange can be (effectively) quenched by dropping pH to ~2.5, and temperature to 0°C
- Exchange rates reveal solvent accessibility



• Problems centre on localisation, and interpretation of solvent accessibility

Labelling technology - Example



- Hydrogen/deuterium exchange of oligomeric 'molecular chaperone' protein
- Side-chains at interfaces exchange relatively rapidly, suggesting a labile oligomer

'Soft' Ionisation

- From molecules in solution to ions in vacuum
- Established ionisation techniques resulted in covalent fragmentation of molecules
- Soft ionisation techniques allow ionisation of large ions without their fragmentation
- Current 'mass record' is >100 MDa!

Nobel Prize in Chemistry 2002

"for their development of soft desorption ionisation methods for mass spectrometric analyses of biological macromolecules"



John B. Fenn 1/4 of the prize USA



Koichi Tanaka 1/4 of the prize Japan

ESI mass spectrum of single protein



Electrospray mass spectra show multiple 'charge states' for a 10 kDa protein

Conformational effects on ESI spectra



 Folded state of protein governs its surface area, and number of sites available for protonation

Maintaining noncovalent interactions



- Transfer multi-subunit protein assembly from solution into gas phase
- Requires control of ionisation conditions, and ion transmission

Nano-electrospray ionisation (nESI)



Collisional focussing



 Both axial and radial components of the ions' velocity can be dampened by collisions with background gas

nESI mass spectrum



Mass accuracy



- Additional mass due to adducted solvent molecules and buffer ions
- Number of adducts inversely related to activation

Membrane protein assemblies are also tractable



- Sample transferred into vacuum within detergent micelle
- Activation within the mass spectrometer removes detergent to leave "naked" protein assembly



Benefits of nESI

- Lower sample amounts (flow rate approx 10nL/min, vs 5 µL/min in ESI)
- Can use aqueous buffers and ambient temperatures
- Narrower charge states due to fewer adduction
- Less dissociation of oligomer
- Symmetrical charge state distribution indicative of a single conformation
- Fewer non-specific aggregates



Non-specific associations during ESI



- Probability of there being >1 analyte molecules in 'final' ESI droplet
- Most droplets are empty, occupancy increases with concentration
- Decreased initial droplet size in nESI reduces prevalence of non-specific aggregates

Intact membrane protein machines - Example



Determining binding affinities - Example



Effects of activation in the gas phase



- Activating conditions lead to high quality mass spectra
- Same activation can lead to
 - collapse (i.e. smaller CCS)
 - then unfolding (i.e. larger CCS)



Benesch, J Am Soc Mass Spectrom (2009), 20, 341-8

Collision induced dissociation of protein assemblies



- Dissociation is asymmetric with respect to mass
- · Unfolded, highly charged monomers are removed sequentially

Deconvoluting heterogeneity with CID



- Peak separation is aided by the charge reduction afforded by CID
- Predictable nature of CID allows back calculation of oligomeric distribution

Quantifying stoichiometries



- MS versus size-exclusion chromatography with multi-angle light scattering
- For proteins of similar composition, abundances match solution values

Free energies from MS measurements - Example



Hilton et al, Proc Roy Soc B (2013), 368, 20110405

Protein dynamics

- Proteins are not static structures, but rather undergo fluctuations both at and before equilibrium
- Such 'protein dynamics' are crucial to their function in the cell
- These dynamics can span a wide range of amplitudes and timescales





Quaternary dynamics - Example



Quaternary dynamics - Example



- Exchange proceeds via the movement of dimeric units
- Incorporation is via sequential incorporation of dimers into oligomers
- · Hetero-assembly leads to a wide variety of possible oligomers

Preservation of structure



 It is clear stoichiometry is preserved in the mass spectrometer, but can we probe native structure?

IM-MS spectrum



Obtaining an experimental CCS



- Every feature resolved in m/z has an associated drift time distribution
- Drift time is converted into CCS either directly or via calibration

CCS values from protein structures



- Can approximate CCS as rotationally averaged projected area
- Determine 'theoretical' CCSs from solved protein structures

CCS comparison



- Excellent correlation between theoretical and measured values for globular proteins
- Discrepancy is due to simplicity of 'projection approximation'
- Correlation motivates use of IM measurements in assessing model structures

Using IM-MS to measure conformers - Example

- P-glycoprotein is an low specificity efflux pump which impairs drug delivery
- IM-MS allows the detection of different conformations - outward, inward open, inward closed (left to right)
- Small molecules affect conformational equilibrium





Using IM-MS to filter structures - Example

- Polydisperse oligomeric protein exists in three different stoichiometries
- Based on comparison with homologous proteins likely structures are polyhedral
- Different polyhedral models can be compared to the IMS measurements





Using IM-MS to filter structures - Example

- Compare random rotations of models to TEM class averages
- Lower score is better fit
- Projected area from TEM is conceptually similar to CCS area from IM
- Combination of techniques reveals ensemble structures that







MS across wide range of time and length scales



In-house software to enable quantification



UniDec.chem.ox.ac.uk



DynamiXL







IMPACT.chem.ox.ac.uk

EMnIM.chem.ox.ac.uk

BioBOx

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